

## Drinking Water Quality of Schools in Theni District

A. JEYASEELI, T. KAVI PRIYA, M. R. DELPHINE ROSE<sup>1</sup> AND M. NARAYANAN

*Department of Advanced Zoology & Biotechnology, St. Xavier's College (Autonomous)  
 Palayamkottai 627002, Tamil Nadu, India*

<sup>1</sup>*Department of Zoology, Jayaraj Annapackiam College (Autonomous)  
 Periyakulam, Tamil Nadu, India*

### Abstract

In the present study the physical and chemical parameters such as appearance, color, odor, turbidity, total dissolved solids, electro-conductivity, pH, total alkalinity, total hardness, calcium, magnesium, sodium, potassium, iron, free ammonia, nitrite, nitrate, chloride, fluoride, sulfate, and phosphate of drinking water supplied to schools in Theni district of Tamil Nadu were studied by collecting samples from twelve schools. Study reveals that the physico-chemical parameters are well within permissible limits of ISI, ICMR and WHO.

**Key words :** Theni district, Schools, Water quality.

Water is one of the most important and abundantly available natural commodity. It is an essential constituent of all plants, animals and human beings. There is a direct impact of quality of water used and human health. The utility of poor quality of drinking water may lead to several life threatening diseases. It was observed that around 20 million people are suffering from fluorosis (Prabhavathi et al. 2001, Jeyaseeti et al. 2006). Since the quality of public health depends to a great extent on the quality of drinking water, it is imperative to determine the physical and chemical characteristics of water to monitor the quality of water supply. The available literature has revealed that no scientific investigation was carried out with regard to the qualitative aspects of ground water for drinking purpose of schools of Theni district. Therefore, the present work was made to assess the water quality status of schools of Theni district.

### Methods

The drinking water samples from 12 schools of Theni district of Tamilnadu were collected in good quality polythene bottles following the standard procedure. Sampling was carried out without adding any preservatives in rinsed bottles directly for avoiding any contamination and brought to the laboratory for the estimation of various physico-chemical param-

eters. Monitoring was done in November 2006. The schools are indicated by S<sub>1</sub> (SDA Matric HSS, Periyakulam), S<sub>2</sub> (Govt Boys HSS, Periyakulam), S<sub>3</sub> (Govt Girls HSS, Periyakulam), S<sub>4</sub> (Kallar HS, Endapudupatti), S<sub>5</sub> (Govt HSS, Vadugapatti), S<sub>6</sub> (Govt HSS, Silvarpatti), S<sub>7</sub> (Govt HSS, Jeyamangalam), S<sub>8</sub> (Govt HSS, Devathanapatti), S<sub>9</sub> (Renuga Vidyalaya, Lakshmipuram), S<sub>10</sub> (Govt HSS, Lakshmipuram), S<sub>11</sub> (Govt HSS, Thamaraikulam) and

**Table 1.** Physical parameters of drinking water in schools of Theni district.

Sta- tions	App- ear- ance	Color	Turbi- dity NT units	Total diss- olved solids (mg/l)	Elec- trical conduc- tivity (Micro mho/ cm)
S <sub>1</sub>	Clear	Colorless	1	535	765
S <sub>2</sub>	Clear	Colorless	1	55	75
S <sub>3</sub>	Clear	Colorless	2	55	75
S <sub>4</sub>	Clear	Colorless	1	615	875
S <sub>5</sub>	Clear	Colorless	1	475	680
S <sub>6</sub>	Clear	Colorless	1	950	1360
S <sub>7</sub>	Clear	Colorless	1	505	720
S <sub>8</sub>	Clear	Colorless	1	165	235
S <sub>9</sub>	Clear	Colorless	1	125	180
S <sub>10</sub>	Clear	Colorless	1	125	180
S <sub>11</sub>	Clear	Colorless	1	55	75
S <sub>12</sub>	Clear	Colorless	1	120	170

**Table 2.** Chemical parameters of drinking water in schools of Theni district. All values are expressed in mg/l except pH.

Stations	pH	Total Alk	Total hardness	Calcium	Magne-sium	Sodium	Pota-ssium	Iron
S <sub>1</sub>	7.7	208	220	44	26	50	15	0.1
S <sub>2</sub>	7.23	20	30	8	2	3	1	0.1
S <sub>3</sub>	7.33	20	30	8	2	3	1	0.1
S <sub>4</sub>	7.6	320	340	68	41	40	2	0.3
S <sub>5</sub>	7.65	280	300	60	36	20	1	0.2
S <sub>6</sub>	7.91	200	500	100	60	80	10	0.1
S <sub>7</sub>	7.92	224	240	48	29	40	5	0.2
S <sub>8</sub>	7.75	76	80	16	10	15	2	0.1
S <sub>9</sub>	7.64	50	60	12	7	10	1	0.3
S <sub>10</sub>	7.69	50	60	12	7	10	1	0.2
S <sub>11</sub>	7.3	20	30	8	2	3	1	0.1
S <sub>12</sub>	7.36	50	60	12	7	6	3	0.2

**Table 2.** Continued.

Stations	Ammonia	Nitrite	Nitrate	Chloride	Fluoride	Sulfate	Phoso-phate	Tidy's Test
S <sub>1</sub>	0.06	0.05	12	100	0.4	14	0.04	0.42
S <sub>2</sub>	0.06	0.01	1	10	0.2	1	0.04	0.21
S <sub>3</sub>	0.06	0.01	1	10	0.2	1	0.04	0.21
S <sub>4</sub>	0.06	0.08	15	70	0.4	20	0.2	0.42
S <sub>5</sub>	0.1	0.06	14	50	0.2	10	0.04	0.63
S <sub>6</sub>	0.09	0.05	20	280	0.4	40	0.04	0.42
S <sub>7</sub>	0.06	0.08	6	60	0.4	12	0.2	0.42
S <sub>8</sub>	0.1	0.08	2	20	0.2	4	0.08	0.42
S <sub>9</sub>	0.06	0.04	2	20	0.2	2	0.04	0.42
S <sub>10</sub>	0.06	0.01	2	20	0.2	2	0.04	0.21
S <sub>11</sub>	0.06	0.01	1	10	0.2	1	0.04	0.21
S <sub>12</sub>	0.1	0.01	2	20	0.2	2	0.08	0.42

S<sub>12</sub> (Govt HSS, Vaigai Dam). The procedures were followed following the standard methods (APHA et al. 1992).

### Results and Discussion

Data on physical and chemical analysis of twelve schools of Theni district are given in Tables 1 and 2 and the standard values in Table 3. The appearance and color of the selected school water samples were clear and colorless. The turbidity is within the permissible limit and ranged from 1 to 2 NT units. The values of total dissolved solids and electro-conductivity varied from 55–950 mg/liter and 75–1,360 mmho/cm respectively. The pH of the selected schools water samples ranged from 7.23 to 7.92 which is well within the prescribed limit of 6.5 to 8.5 of the ISI standards (WHO 1992).

The total alkalinity of all the collected samples ranged from 20–30 mg/liter. Samples S<sub>1</sub>, S<sub>4</sub>, S<sub>5</sub> and S<sub>7</sub> have alkalinity values beyond the permissible limit of ISI (150–200 mg/liter). Alkalinity alone is not harmful to human beings (Pande and Sharma 1999). Excess alkalinity gives a bitter taste to water. Total hardness ranged from 30–500 mg/liter and in stations S<sub>1</sub>, S<sub>4</sub>, S<sub>5</sub>, S<sub>6</sub> and S<sub>7</sub> the total hardness above the permissible limit (P=100 mg/liter, WHO 1992) and it is important in determining the suitability of a water for domestic and industrial uses.

Calcium values are varied from 8–100 mg/liter and these values are within permissible limits as prescribed by WHO (1992). In human body it is essential for muscular and nervous system, cardiac functions and in coagulation of blood. Therefore, low level calcium may have adverse effect on human health (Sharma and Singh Chandel 2005). In this observa-

**Table 3.** Standard physico-chemical values for drinking water. All values are expressed in mg/l except turbidity, EC and pH. P=Permissible limit ; E=Excessive limit.

Parameters	ISI		ICMR		WHO	
	P	E	P	E	P	E
<b>Physical</b>						
Color	10	50	5	25	5	25
Taste & Odor	Unobject	–	Unobject	–	Unobject	–
Turbidity	10	25	5	25	5	25
Electro-conductivity	–	–	–	–	500	1500
<b>Chemical</b>						
pH	6.5–8.5	6.5–9.2	7–8.5	6.5–9.2	7–8.5	6.5–9.2
TDS	–	–	–	–	500	1500
Alkalinity	–	–	–	–	120	–
Total hardness	300	600	300	600	–	–
Calcium	75	200	75	200	75	200
Iron	0.3	1.0	0.3	1.0	0.3	1.0
Chlorides	250	1000	250	1000	200	600
Sulfates	150	400	200	400	200	400
Nitrates	45	–	20	50	–	100
Fluorides	0.6–1.2	–	1.0	2.0	0.5	1.0–1.5

tion, magnesium values ranged from 2–60 mg/liter and these are well within the desirable limit. Magnesium rich water may cause the gastro-intestinal irritation in the presence of sulfate containing large amount of  $Mg-SO_3$  may be purgative and cause diarrhoea amongst the consumers.

Sodium and potassium values ranged from 3–80 and 1–15 mg/liter respectively. The presence of sodium is an important parameter the excess quantity of which can deteriorate the soils. The iron content ranged from 0.1–0.3 mg/liter and free ammonia from 0.06–0.1 mg/liter. Iron is considered as a nuisance in excessive quantities. Nitrite and nitrate values were between 0.01–0.8 and 1–20 mg/liter. The nitrate values are within the permissible limit. Nitrate is dangerous for human health especially in infant below six months of age.

Chloride values varied from 10–280 mg/liter and are within permissible limits as prescribed by WHO except  $S_6$  (permissible 200 mg/liter). In reasonable concentrations these are not harmful to human beings and imparts salty taste beyond the concentration of 250 mg/liter. The fluoride content varied from 0.2–0.4 mg/liter. The presence of large amount of

fluoride is associated with skeletal and dental fluorosis. Sulfate values varied between 1–40 mg/liter and within permissible limits. If the concentration is more than 500 mg/liter, it causes diarrhoea in human beings and cattle. Tidy's values ranged from 0.21–0.63 mg/liter.

The physico-chemical analysis of drinking water samples in schools of Theni district reveals that all the parameters are well within the permissible limit of WHO. However, monitoring is important to detect drinking water contamination and to provide an advanced warning to the consumers. The results of suitability evaluation denotes that there is no major hazard in selected drinking water samples in schools of Theni district of Tamilnadu.

#### References

- APHA, AWWA and WPCF. 1992. *Standard methods for the examination of water and wastewater*. 18th ed. Am. Pub. Hlth. Assoc., Washington DC, USA.
- Jeyaseeli A., K. S. Priya, M. R. Delphine Rose, S. Nirmala and M. Narayanan. 2006. Critical analysis of ground water quality of Nilakkottai Taluk of Tamil Nadu. *Indian J. Curr. Sci.* 9 : 931–934.
- Pande K. S. and S. D. Sharma. 1999. Studies on water

- quality index for Ramganga river at Moradabad, Uttar Pradesh. *Poll. Res.* 18 : 327—333.
- Prabhavathi N., B. Kavita, D. A. Jeyakar Chellaraj and R. E. Raja. 2001. Analysis of fluorida in ground and surface water and survey of dental fluosis among school children in Lalgudi Taluk, Trichy. *Indian J. Environ. Prot.* 21 : 51—53.
- Sharma S. K. and C. P. Singh Chandel. 2005. A study on ground water pollution of Amer and Barath blocks of Jaipur district (Rajasthan). *Aguacult.* 6 : 197—204.
- World Health Organizations. 1992. *International standard for drinking water*. WHO, Geneva, Switzerland.