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Effect of Soil and Foliar Application of Zinc on Quality Parameters of Cauliflower (*Brassica oleracea* var *botrytis* L.)

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ABSTRACT

A field experiment on Effect of soil and foliar application of zinc on growth and yield parameters of cauliflower (*Brassica oleracea* var *botrytis* L.) was conducted in the farmer's field during *kharif* 2018. The experiment was laid out in a Randomized Complete Block Design with 9 treatments and 3 replications. The results revealed that application of 4 kg of Zn ha⁻¹ through ZnSO₄ as soil application + 0.5% Zn through zinc sulfate as foliar spray along with RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) + 25 t of FYM per hectare significantly increases the quality

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B. S. Lalitha Department of Agronomy, College of Agriculture, GKVK, UAS, Bengaluru 560065, India email : chethanakh91@gmail.com *Corresponding author parameters viz., total soluble solids and ascorbic acid (vitamin C) were increased significantly with soil and foliar application of zinc.

Keywords Quality parameters, Ascorbic acid, Cauliflower, Zinc sulfate, Foliar application.

INTRODUCTION

Among micronutrients, zinc is considered as one of the first nutrient essential for the plants and has got greater significance in crop production due to its wider occurrence of deficiency in different agro-climatic regions of the country and spectacular response of crops to its application. The extent of Zn deficiency varies widely with the nature of soil. Coarse textured, calcareous or alkaline pH, low organic carbon and highly eroded soils are more prone to occurrence of zinc deficiency. In India more than 40% of soils are deficient in zinc and in Karnataka 73% of soils are deficient in zinc (Srinivasarao and Sudha 2013). Cauliflower (Brassica oleracea var botrytis L.) is one of the important cole crops grown in India which belongs to family Brassicaceae. Cauliflower is originated in ancient Asia minor in northeastern part of the Mediterranean. It is one of the important daily con-

sumed vegetable in the world. Cauliflower is grown extensively in tropical and temperate regions of the world viz., China, Germany, India, Indonesia, Japan, Korea, Poland, Russia, Taiwan, Turkey, USA and several other countries (Rai and Yadav 2005). Cauliflower is known for its nutritive value, digestibility and for preparation of delicious dishes. Cauliflower seedlings are used for salad and green. The curd is used in curries, soups and pickles. It contains higher amount of vitamin C, folate and also rich source of vitamins like riboflavin, thiamine, nicotinic acid and has antioxidants. It contains indole-3-cardinal that affects the metabolism of estrogen. The extract from the inflorescence used to cure scurvy. It decreases the risk of heart attack, brain disorder, reduces the blood cholesterol and also reduces the risk of cancer. It lowers the blood cholesterol and also good for skin health. Deficiency of Zn noticed in most soils of Karnataka. Though some work has been done on the influence of these micronutrients in field and fruits crops (Babu and Singh 2001 in Litchi) not much information is reported regarding their effect on vegetables especially cole crops. The literature pertaining to zinc nutrition on cole crops is very scanty. Keeping in view of the above facts a research work entitled Effect of soil and foliar application of zinc on quality parameters of cauliflower (*Brassica oleracea* var *botrytis* L.).

MATERIALS AND METHODS

Location of the experimental site

The experiment was carried out in the farmer's field at Konapalli, Chintamani Taluk, Chickkaballapur District. The experimental farm is situated in Eastern Dry Zone (Zone 5) of Karnataka at 13.4020° N latitude and 78.0551° E longitude with elevation of 865 m above mean sea level. Initial physical and chemical parameters of soil given in Table1.

Detailed program of work

Fertilizer application

Recommended dose of fertilizers for cauliflower crop is 150 : 100 : 125 kg N, P_2O_5 , K_2O ha⁻¹ and FYM 25 t ha⁻¹ and they were applied according to the treatment details. Nitrogen in the form of urea, P_2O_5 in the form of SSP, K_2O in the form of muriate of potash and zinc in the form of zinc sulfate were applied.

 Table 1. Initial physical and chemical properties of soil at experimental site.

| Sl. No. | Soil property | Content | Method followed |
|---------|--|---------|---|
| | Sand (%) | 57.84 | |
| | Silt (%) | 7.69 | |
| 1. | Clay (%) | 34.47 | International pipette method |
| | | Sandy | (Piper 1966) |
| | Textural class | Clay | |
| 2. | pH (1 : 2.5) | 8.36 | Potentiometry (Jackson 1973) |
| 3. | EC (1 : 2.5) (dS m ⁻¹) | 0.22 | Conductometry (Jackson 1973) |
| 4. | OC (%) | 0.63 | Wet oxidation method (Jackson 1973) |
| 5. | Available N (kg ha ⁻¹) | 382.60 | Alkaline permanganate method |
| | | | (Subbaiah and Asija 1956) |
| 6. | Available P_2O_5 (kg ha ⁻¹) | 32.30 | Oleson's method (Jackson 1973) |
| | 2 0 | | Neutral IN ammonium acetate |
| 7. | Available K_2O (kg ha ⁻¹) | 208.00 | extraction & flame photometry method |
| | - | | (Page et al. 1982) |
| 8. | Exchangeable Ca (cmol (p ⁺) kg ⁻¹) | 6.75 | Versenate titration method (Jackson 1973) |
| 9. | Exchangeable Mg (cmol (p ⁺) kg ⁻¹) | 3.0 | Versenate titration method (Jackson 1973) |
| 10. | Available S (mg kg ⁻¹) | 16.6 | Turbidometry method (Black 1965) |
| 11. | DTPA extractable Zn (mg kg ⁻¹) | 0.46 | DTPA extraction, atomic absorption |
| 12. | DTPA extractable Fe (mg kg ⁻¹) | 1.98 | spectrophotometer method (Lindsay |
| 13. | DTPA extractable Mn (mg kg ⁻¹) | 1.95 | and Norvel 1978) |
| 14. | DTPA extractable Cu (mg kg ⁻¹) | 0.42 | |
| 15. | Available B (mg kg ⁻¹) | 0.34 | Hot water soluble extraction method |
| | | | (Berger and Troug 1939) |

Foliar spray: As per the treatment details, the foliar spray of zinc @ 0.5% through zinc sulfate was taken at 30 days after transplanting.

Quality parameters

Chlorophyll content (SPAD meter reading) : By using SPAD-502 meterchlorophyll content of randomly selected plants were recorded at 45 days after transplanting and at harvest.

Ascorbic acid content (vitamin C) : Ascorbic acid content was estimated by the procedure outlined by AOAC (1970) which is based on the reduction of 2, 6-dichlorophenol indophenol (2, 6-DCPIP) by ascorbic acid.

Extraction: A tissue sample of 1 g was macerated with 4 ml of 3% metaphosphoric acid in a mortar and pestle. The homogenate was centrifuged for 20 minutes at1000 rpm and then the supernatant was carefully decanted into a flask and final volume was made up to 25 ml with 3% metaphosphoric acid.

Estimation: An aliquot sample of the extract was titrated with 2, 6-dichlorophenol indophenol reagent until a pink end-point (which persists for 15 seconds). A standard curve was prepared by titrating a known amount of ascorbate (1-50 mg) with 2, 6-dichlorophenol indophenols reagent. Total amount of ascorbate present in the sample was calculated from the standard

Table 2. The details of field experiment conducted are given in Table 2.

| Test crop | : Cauliflower | | |
|--|---|--|--|
| | | | |
| Variety | : Unnathi | | |
| Spacing | $: 45 \text{ cm} \times 30 \text{ cm}$ | | |
| Location | : Farmer's field, Chintamani, | | |
| | Eastern dry zone | | |
| Design | : RCBD | | |
| Plot size | $: 3.6 \times 3 \text{ m} (10.8 \text{ m}^2)$ | | |
| No. of treatments | : 9 | | |
| No. of replications | : 3 | | |
| Source of zinc | : Zinc sulfate | | |
| Foliar spray | : 30 DAT | | |
| Recommended dose of | | | |
| fertilizers | : 150 : 100 : 125 | | |
| $(N : P_2O_5 : K_2O \text{ kg ha}^{-1})$ | | | |
| FYM $(t ha^{-1})^2$ | : 25 | | |

| Table | 2 | Continued. |
|-------|----|------------|
| rance | ₩. | Commucu. |

| | Treatment details |
|---|---|
| Sl. No | . Treatment details |
| T, | RDF + FYM |
| T_2^1 | $T_1 + 2$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application |
| Τ, | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application |
| T ₄ | $T_1 + 6$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application |
| T ₅ | $T_1 + 8$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application |
| T ₆ | $T_1 + 1$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application |
| | + 0.5% Zn through ZnSO ₄ as foliar spray |
| T ₇ | $T_1 + 2$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil applica- |
| | tion + 0.5% Zn through $ZnSO_4$ as foliar spray |
| T ₈ | $T_1 + 3$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil applica- |
| | tion + 0.5 % Zn through $ZnSO_4$ as foliar spray |
| T ₉ | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil applica- |
| , i i i i i i i i i i i i i i i i i i i | tion+ 0.5% Zn through $ZnSO_4$ as foliar spray |

curve. The results were expressed in mg ascorbic acid per 100 g fresh weight. Ascorbic acid was calculated by using following formula :

| Ascorbic acid = (Titer value×Dye factor×Volu | ume made up) |
|--|--------------|
| | ×100 |

(mg 100 g⁻¹) (Volume of filtrate taken×Wt. of sample taken)

Total soluble solids (TSS) : Total soluble solids were estimated by using Erma Hand Refractometer and value was recorded as ^oBrix at room temperature.

RESULTS AND DISCUSSION

Effect of soil and foliar application of zinc on chlorophyll content (SPAD meter reading) of cauliflower

Data on chlorophyll content at growth different stages of cauliflower as influenced by soil and foliar application of zinc are presented in Table 3. Chlorophyll content differed significantly with the soil and foliar application of zinc. At 45 DAT, treatment which received 4 kg of Zn ha⁻¹ through ZnSO₄ as soil application + 0.5% Zn through ZnSO₄ as foliar spray and RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) along with FYM (T₉) has recorded maximum chlorophyll content (57.88) followed by treatment (T₅) which received T₁ + 8 kg of Zn ha⁻¹ through ZnSO₄ as soil application (56.91) and T₁ + 3 kg of Zn ha⁻¹ through ZnSO₄ as foliar spray (T₈) which has recorded 56.27. Lowest was recorded inT₁ treatment (51.78). At harvest, significantly high-

| | Treatments | Chlorophyll content | | |
|--|--|---------------------|------------|--|
| | | 45 DAT | At harvest | |
| T ₁ | RDF + FYM (150 : 100 : 125 N - P_2O_5 -K ₂ O kg ha ⁻¹ + 25 tons FYM ha ⁻¹) | 51.78 | 54.27 | |
| Γ, | $T_1 + 2$ kg of Zn ha ⁻¹ through ZNSO ₄ as soil application | 52.64 | 54.92 | |
| Γ_2 Γ_3 Γ_4 Γ_5 Γ_6 | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application | 53.34 | 54.66 | |
| ٢́, | $T_1 + 6$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application | 55.12 | 56.40 | |
| Γ | $T_1 + 8 \text{ kg of } Zn \text{ ha}^{-1}$ through $ZnSO_4$ as soil application | 56.91 | 58.37 | |
| r, | $T_1 + 1$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | | |
| 0 | $ZnSO_4$ as foliar spray | 52.51 | 55.26 | |
| 7 | $T_1 + 2$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | | |
| / | $ZnSO_4$ as foliar spray | 53.85 | 56.59 | |
| Г ₈ | $T_1 + 3$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | | |
| 0 | $ZnSO_4$ as foliar spray | 56.27 | 59.28 | |
| Γ. | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | | |
| , | $ZnSO_4$ as foliar spray | 57.88 | 60.09 | |
| Em ± | * * * | 1.268 | 1.271 | |
| CD (5%) | | 3.803 | 3.812 | |

| Table 3 | Effect of soil and fol | iar application | of zine on chloro | nhvll content | (SPAD reading) of cauliflow | er |
|----------|------------------------|-----------------|-------------------|----------------|-----------------------------|-----|
| Table 5. | Effect of son and to | iai application | of zine on emore | phyn content i | (SIAD reading) of caulinow | UI. |

er chlorophyll content (60.09) was recorded in the treatment received 4 kg of Zn ha⁻¹ through ZnSO₄ as soil application + 0.5% Zn through ZnSO₄ as foliar spray and RDF (150 : 100 : 125 N-P₂O₅ –K₂O kg ha⁻¹) along with FYM (T₉) followed by T₁ + 3 kg of Zn ha⁻¹ through ZnSO₄ as soil application + 0.5% Zn through ZnSO₄ as foliar spray (T₈) which has recorded 59.28 and T₁ + 8 kg of Zn ha⁻¹ through ZnSO₄ as soil application (T₅) which has recorded 58.37. Lowest value was recorded in the treatment received RDF + FYM (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹ + 25 tons FYM ha⁻¹) which has recorded 54.27.

The catalytic or stimulatory effect of zinc on most of the physiological and metabolic process of plant may be responsible for the growth of plant. Zinc is also an essential component of enzymes and plays an important role in nitrogen metabolism which resulting in uptake of nitrogen by the plants. The synergistic effect of zinc on nitrogen resulted in increased uptake of nitrogen by the plants and hence there is an increased growth. Zinc is necessary for normal cell division and other plant metabolic processes (Singh and Verma 1991), it is helpful in improving the growth of cauliflower by increasing the growth promoting hormone i.e., auxin content of plant and playing a role in cellular oxidation. Increased chlorophyll content might be due to fact that besides the role of zinc in chlorophyll formation, it also influenced cell division, meristematic activity of tissues and expansion of cell and formation of cell wall. Similar results were reported by Lashkari et al. (2008). Sivakumar et al. (2005) also reported that the foliar application of zinc @ 0.5% resulted in highest chlorophyll content in okra.

Effect of soil and foliar application of zinc on TSS and ascorbic acid content of cauliflower

The quality parameters like total soluble solids (°Brix) and ascorbic acid (mg 100 g⁻¹) as influenced by soil and foliar application of zinc is presented in Table 4.

Table 4. Effect of soil and foliar application of zinc on quality parameters of cauliflower.

| | Treatments | Total soluble solids (°Brix) | Ascorbic acid (mg 100 g ⁻¹) |
|---------|---|---------------------------------|--|
| T. | RDF + FYM (150 : 100 : 125 N-P ₂ O ₅ -K ₂ O kg ha ⁻¹ + 25 tons FYM ha ⁻¹) | 5.10 | 49.76 |
| Τ, | $T_1 + 2$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application | 5.47 | 50.76 |
| T, | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application | 5.40 | 52.19 |
| T, | $T_1 + 6$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application | 5.70 | 55.43 |
| T_{5} | $T_1 + 8 \text{ kg of Zn ha}^{-1}$ through $ZnSO_4$ as soil application | 6.00 | 59.64 |

Table 4. Continued.

| | Treatments | Total soluble solids (°Brix) | Ascorbic acid (mg 100 g ⁻¹) |
|----------------|---|---------------------------------|--|
| Г, | $T_1 + 1$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | |
| 0 | $ZnSO_4$ as foliar spray | 5.33 | 51.32 |
| Γ ₇ | $T_1 + 2$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | |
| , | $ZnSO_{A}$ as foliar spray | 5.37 | 54.34 |
| F.8 | $T_1 + 3$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | |
| 0 | $ZnSO_{4}$ as foliar spray | 6.10 | 57.54 |
| 9 | $T_1 + 4$ kg of Zn ha ⁻¹ through ZnSO ₄ as soil application + 0.5% Zn through | | |
| 7 | $ZnSO_4$ as foliar spray | 6.13 | 60.13 |
| | SEm ± | 0.162 | 1.350 |
| | CD (5%) | 0.487 | 4.049 |

Total soluble solids ("Brix) : The soil and foliar application of zinc significantly influenced the total soluble solids in cauliflower curd. Significantly higher total soluble solids (6.13 °Brix) was recorded with the treatment combination of 4 kg of zinc ha-1 as soil application + 0.5% zinc as foliar spray along with RDF $(150:100:125\,\text{N-P}_{2}\text{O}_{5}\text{-}\text{K}_{2}\text{O}\,\text{kg}\,\text{ha}^{\text{-1}})$ and FYM (T_{o}). It was found at par with T_5 which received 8 kg of zinc ha⁻¹ as soil application along with RDF + FYM and T_o which received 3 kg of zinc ha⁻¹ as soil application + 0.5% zinc as foliar spray along with RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) and FYM. Lowest total soluble solids (5.10 °Brix) was recorded in T₁ which received RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) and FYM. Earlier it was reported that significantly higher TSS in the cauliflower was observed with application of 40 ppm zinc as foliar spray.

Ascorbic acid (mg kg⁻¹): Significantly higher ascorbic acid (60.13 mg 100 g⁻¹) was recorded in T₉ (T₁ + 4 kg of zinc ha⁻¹ as soil application + 0.5% zinc as foliar spray) followed by T₅ which received 8 kg of zinc as soil application along with RDF + FYM which has recorded 59.64 mg 100 g⁻¹. Lowest ascorbic acid (49.76 mg 100 g⁻¹) was recorded in the treatment T₁ which received only RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) and FYM.

Total soluble solids and ascorbic acid content of cauliflower were significantly influenced by soil and foliar application of zinc. Application of zinc @ 4 kg ha⁻¹⁺ 0.5% zinc as foliar spray along with recommended dose of fertilizers (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) and FYM resulted in increased quality attributes like TSS and ascorbic acid in cauliflower.

Similar results were obtained by Verma et al. (2017), stated that application of zinc at 5-7.5 kg ha⁻¹ recorded significantly higher protein and ascorbic acid content in cauliflower curd. Application of 30 kg zinc ha⁻¹ resulted in maximum and significantly increased the quality attributes viz., total soluble solids, ascorbic acid, total sugars, reducing sugars and non-reducing sugars in sprouting broccoli (Shivran et al. 2017).

Zinc played a major role in increasing CEC of roots which helps in increasing absorption of nutrient from the soil. Further, it helps in synthesis of chlorophyll and maintaining the auxin concentration and its stimulatory effect on plant physiological and metabolic process, might have helped the plants in increasing the quality attributes. The results were found in line with Gajendra et al. (2014), stated that TSS and ascorbic acid content has increased by application of zinc along with boron and manganese in broccoli. Similar results were obtained by Kotecha et al. (2016), reported that the application of zinc sulfate at 0.5% ha significantly increased the ascorbic acid in cabbage. Results were in line with the findings of Kaur et al. (2018) in potato, Verma et al. (2017) in cauliflower, Zhao (2006) in cabbage, Ramachandra et al. (2014) in tomato and Singh et al. (2018) in potato.

CONCLUSION

Quality parameters viz., total soluble solids and ascorbic acid (vitamin C) were increased significantly with soil and foliar application of zinc. Application of 4 kg of Zn ha⁻¹ through ZnSO₄ as soil application + 0.5% Zn through zinc sulfate as foliar spray along with RDF (150 : 100 : 125 N-P₂O₅-K₂O kg ha⁻¹) and

FYM (25 t ha⁻¹) significantly increased the total soluble solids (6.13 °Brix) and ascorbic acid (vitamin C) which recorded 60.13 mg 100 g⁻¹.

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